Method Implementation Document for EN 14385

BS EN 14385:2004

Stationary source emissions – Determination of the total emission of As, Cd, Cr, Co, Cu, Mn, Ni, Pb, Sb, Tl and V

Measurement of metals including an option to measure mercury

Environment Agency
Version 4
December 2013
## Record of amendments

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### Status of this document

This method implementation document may be subject to review and amendment following publication. The most recent version of this MID can be accessed from:

[www.mcerts.net](http://www.mcerts.net)

It is also available on the Source Testing Association web site at:

[www.S_T_A.org](http://www.S_T_A.org)

### Implementation date

It is expected that organisations, which hold MCERTS accreditation for BS EN 14385, will have met the requirements of this version by 1 July 2014.

### Feedback

Any comments or suggested improvements to this MID should be e-mailed to Rupert Standring at rupert.standring@environment-agency.gov.uk.
Role of Method Implementation Documents

The Environment Agency recognises that European and International standards may need supplementing by Method Implementation Documents (MIDs) to ensure they are being implemented consistently.

We have established our Monitoring Certification Scheme: MCERTS to deliver quality environmental measurements. Organisations wishing to include a standard in their schedule of MCERTS accreditation shall follow the requirements of the standard and, where available, the associated MID.

It may not be necessary to produce a MID for every standard but where required they will be used to supplement standards called up by Technical Guidance Note M2. MIDs provide details on how the preferred standards shall be used for regulatory monitoring.

MIDs are produced in collaboration with the Source Testing Association and its members.

Copies of MIDs and further information on MCERTS including copies of performance standards and guidance can be obtained from our web-site at:

www.mcerts.net

Or from the STA web-site at:

www.S-T-A.org
Introduction

This document supplements BS EN 14385:2004 (referred to in this document as EN 14385). However, it does not re-state all the provisions of EN 14385 and organisations are reminded of the need to comply with the criteria detailed in EN 14385. The clause numbers in this document follow those of EN 14385 although the text from the standard is not repeated in this document. EN 14385 remains the authoritative document and in cases of dispute, the accreditation body will adjudicate on unresolved matters.

1. Scope

European Standard BS EN 14385:2004 may be used to determine other metals than those named in the scope of the standard.

In order to extend the scope of metals it is necessary for the analytical laboratory to carry out validation for the analysis of the additional metals. The filter, rinsing solutions and absorption solutions can be spiked with certified reference materials and analysed to determine recoveries and limits of detection.¹

This MID includes two approaches. One for measuring metals including mercury and one for measuring metals without mercury.

EN 14385 allows BS EN 13211 - Manual method of the determination of the concentration of total mercury (referred to in this document as EN 13211) to be carried out using a side stream sampling technique. In the UK custom and practice does not normally include the use of side stream sampling techniques.

This MID therefore includes the option of measuring mercury in the main sampling train. When carrying out this option it is important that all provisions specific to EN 13211 are met.

Measurement of metals including mercury differs from measurement of metals, according to EN 14385, in the following ways:

- a different absorber (also known as impinger) train configuration to that specified by EN 14385 is used;
- the absorbants / analytical requirements of BS EN 13211 are included; and
- extra analytical requirements are employed to ensure the combination of both methods does not affect the reliability of the measurement.

This approach follows the principles established by US EPA Method 29 and a validation exercise carried out by the Environment Agency.²

In order to obtain MCERTS accreditation for measuring metals including mercury it is necessary to meet the requirements of this MID and be accredited to use EN 13211. If accreditation is sought for metals excluding mercury it is not necessary to be accredited for EN 13211. The UKAS schedule of accreditation will specify whether mercury is included or excluded.

¹ The analysis of tin (Sn) may be problematic due to the formation of "tin stone" (a mixed valence oxide). However, validation work carried out by a UK laboratory has shown acceptable results.

² Agency document: Monitoring mercury at the same time as other heavy metals (2004)
Furthermore, it is necessary for the analytical laboratory to be accredited for the analysis specified in the standards and in this MID. This should be stated on the analytical laboratory’s schedule of accreditation.

EN 14385 has been validated on stacks with a gas composition given in Table 1 of the standard. The standard can be used on processes with other gas compositions.

Note 1: practical experience shows that the standard can be used on a wide range of processes and stack gas conditions.

Note 2: expected concentrations and conditions of the stack gas are noted in the site specific protocol at the measurement planning stage.

It is not acceptable to combine the measurement of metals and particulates (see section 8.7 of EN 14385 and section 7.5 and 9 of BS EN 13284-1).

2. Normative references
MID 13284-1, Environment Agency.
USEPA Method 29 *Determination of metal emissions from stationary sources* should be referred to if mercury is being measured.

3. Definitions, symbols and abbreviations
3.1 Definitions
No additional information.

3.2 Symbols
No additional information

3.3 Abbreviations
No additional information

4. Principle
No additional information

5. Apparatus, chemicals and gases
5.1 Apparatus
5.1.1 General
Commercial titanium is made to different specifications based on the purity of the titanium. As the other materials present may contain the specific elements being determined during the measurement it is important that an appropriate grade of titanium is used.

Titanium will degrade in gas streams with a fluoride content of greater than 20ppm.

Titanium may also degrade at temperatures greater than 350°C.

When using this MID to sample for metals including mercury it is important that parts of the sampling equipment in contact with the mercury containing flue gas, or with liquid containing mercury, shall be made of materials shown in Table 1 of EN 13211.

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5.1.2 Apparatus for sampling

5.1.2.1 Nozzle

BS EN13284-1 states that nozzles with an internal diameter of less than 6mm shall be avoided. At high stack gas velocities (typically above 25 m/s) it may be necessary to use a nozzle with a diameter of less than 6mm in order to remain within the designed flow rate of the sampling train.

If a nozzle with a smaller diameter than 6mm is used:
- sufficient sample volume shall be extracted to meet the limit of detection required by the analytical laboratory;
- justification for the deviation from EN 13284-1 shall be provided in the monitoring report.

5.1.2.2 Filter housing and filter support

During the production of a site-specific protocol it is important to specify whether an in or out of stack filter will be used.

In the UK it is general custom and practice to use an out of stack filter system to measure metals. Annex B of this document shows the sample train with an out of stack filter configuration.

5.1.2.3 Sampling probe

No additional requirements

5.1.2.4 Absorbers

Absorbers shall be kept below 30°C. This can be done by using ice.

5.1.2.5 Connection fittings and tubing

PTFE is not suitable for temperatures in excess of 200°C

5.1.2.6 Suction unit

No additional requirements

5.1.2.7 Gas volume metering

No additional requirements

5.1.2.8 Storage bottles

No additional requirements

5.1.3 Apparatus for analysis

5.1.3.1 Digestion device

No additional requirements

5.1.3.2 Digestion PTFE vessels

No additional requirements

5.1.3.3 Measuring flasks

No additional requirements

5.1.3.4 Storage flasks

No additional requirements

5.1.3.5 Pipettes and dispensers

No additional requirements
5.2 Chemicals and filter material

5.2.1 General
No additional requirements

5.2.2 Filter
A batch of filters for metals sampling are those prepared on a single occasion. This may include filters prepared prior to the sampling campaign for use exclusively on that campaign or a greater number of filters prepared on one occasion but stored for use in future monitoring campaigns.

At least one filter per batch should be analysed for metal content prior to use.

5.2.3 Hydrofluoric acid
No additional requirements

5.2.4 Hydrogen peroxide
No additional requirements

5.2.5 Nitric acid
No additional requirements

5.2.6 Hydrochloric acid
No additional requirements

5.2.7 Water
No additional requirements

5.2.8 Absorption solution
\[\text{EN 14385 states that the absorption solution must be freshly made. The diluted nitric acid solution shall be prepared at a permanent laboratory. The addition of peroxide to the acid solution may be carried out at the laboratory or on site.}\]

The elements being determined shall be below 1μg/l in the absorption solution.

\[\text{EN 13211 describes two absorption solutions that may be used for the absorption of mercury.}\]

If a potassium permanganate / sulphuric acid solution is used it shall be prepared at a permanent laboratory, as it requires the addition of concentrated hydrochloric acid.

If a potassium dichromate / nitric acid solution is used the nitric acid solution shall be made in a permanent laboratory. The potassium dichromate solid can be added to the pre-prepared nitric acid solution at the laboratory or on site.

\[\text{If a potassium permanganate / sulphuric acid solution is used it is important that the solution is not mixed with the absorption solutions for other gaseous metals. Mixing of the two may lead to the nitric and sulphuric acid causing a gas release, which results in high pressure and possible explosion of the bottle. The use of an empty absorber between the different absorption solutions will reduce the likelihood of mixing occurring (see Annex B).}\]

5.2.9 Boric acid solution
No additional requirements
5.2.10 Chemical blank solution for digestion
At least one digestion blank shall be prepared with each batch of samples.

5.2.11 Rinsing acid
The standard states that the rinsing acid is used for rinsing the digestion equipment. The rinsing acid is used on site to rinse the nozzle probe and filter housing (see section 8.4.2)

EN 14385 specifies the use of 25% mass content nitric acid for rinsing glassware on site. Due to health and safety reasons 5% mass content nitric acid may be used.

The elements being determined shall be below 1μg/l in this solution. This shall be confirmed before use.

5.2.12 Standard solutions
No additional requirements

5.2.13 Certified reference material
No additional requirements

5.2.14 Gases
No additional requirements

6. Sampling equipment
No additional requirements

6.1 General requirements
When performing the determination of mercury in conjunction with EN 14385 the first three absorbers in the sample train are the same as specified in EN 14385. The fourth absorber is empty. The fifth and sixth absorber contains absorbent solutions for gas phase mercury. It may be necessary to place an empty absorber at the start of the train and before the drier assembly (the use of these optional absorbers depends on the expected moisture content of the stack gas). An example sample train is provided in Annex B.

6.2 Isokinetic sampling equipment
No additional requirements

6.3 Absorbers
The absorber efficiency test compares the concentration of each element found in the third absorber with the concentration of that element in the total sample train.

Note 1: A number of metals are often measured together as a group and are summed for comparison with an ELV for total metals. For example, metals are typically reported in two groups:

- Cd and Tl
- Sb, As, Pb, Cr, Co, Cu, Mn, Ni, & V

The absorber efficiency test is applied separately to each group of metals.

If one of the individual metal elements in the last impinger (e.g. Sb) is more than 10% of the total combined metals for that individual element (i.e. the filter, probe rinse and impingers) collected, then the result for all the metals in that group shall be rejected.
However, for each group the requirement to pass the absorber efficiency check shall not apply:

- to individual metal element(s), where the individual mass of the element in the final result is less than 1% of the result for the total mass of metals in the group; or
- if the total combined result for the metals in the group is less than 30% of the ELV.

Note 2: the absorber efficiency check becomes less effective as the concentration of the pollutant in a stack gas decreases.

Note 3: For example:
- Cd and Tl combined – 0.05 mg/m³
- Sb, As, Pb, Cr, Co, Cu, Mn, Ni, V combined – 0.5 mg/m³

For Cd and Tl combined it would not be necessary to comply with the less than 10% in the last impinger criteria, if the overall combined Cd and Tl result for the sample train was below 0.015 mg/m³ (i.e. <30% of the ELV).

For combined metals (Sb, As, Pb, Cr, Co, Cu, Mn, Ni, V) it would not be necessary for each individual element to comply with the less than 10% in the last impinger criteria, if the overall combined result for the sample train was below 0.15 mg/m³ (i.e. <30% of the ELV).

For mercury the requirements on the absorber efficiency in EN 13211 shall be followed. However, when EN 14385 and EN 13211 are combined, the mercury in the last absorber is compared to the total mercury in both the mercury absorbers and the absorbers used for metals.

7. Cleaning of the sampling equipment prior to sampling

Rinse all sampling train glassware with hot tap water and then wash in hot soapy water. Next, rinse glassware three times with tap water, followed by three additional rinses with HPLC water. Then soak all glassware in a 10 percent (v/v) nitric acid solution for a minimum of 4 hours, rinse three times with water and then rinse a final time with acetone and allow to air-dry. Cover all glassware openings until the sampling train is assembled for sampling³.

Note: the grade of water is defined in section 5.2.7 of the standard.

The above procedure also includes glass and titanium probe liners. However, it may not be practical to soak probe liners for four hours in a nitric acid solution. An alternative would be to circulate the rinsing solution for about 15 minutes.

Glassware shall be cleaned according to the above procedure prior to a sampling campaign.

Although it is acceptable to re-use glassware on different stacks, it may not be advisable due to the risk of contamination.

8. Procedure

8.1 General requirements

The sampling time shall be related to the LOD of the analysis method and the expected stack gas concentration.

³ This procedure is specified in US EPA method 29
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There are several temperature criteria required depending on the stack gas temperature, the type of equipment being used and whether sampling for mercury is being undertaken.

EN 14385 states that the sampling probe and out-stack filter shall be maintained at a temperature of at least 20°C above stack gas temperature. However, due to limitations on the equipment, it is not necessary to heat the sampling probe and out-stack filter to temperatures above 180°C.

The following applies when sampling for metals including mercury:

- the sampling probe and the out-stack filter shall be maintained at a temperature of at least 20°C above stack gas temperature (it is not necessary to heat the sampling probe and out-stack filter to temperatures above 180°C).
- when using titanium equipment the temperature shall be maintained at 180°C or higher.

Details of the probe and filter temperature shall be included in the report.

When measuring mercury it is important that the absorption solutions do not mix. Mixing of the solutions will result in manganese contamination.

8.2 Preparation and installation of equipment

8.2.1 Filter installation

Filters should not be installed or recovered on exposed platforms.

8.2.2 Nozzle installation

No additional requirements

8.2.3 Installation of absorbers

Due to the possibility of pressure caused by stored absorber solutions, it may be appropriate to consider the use of pressure-relief devices on absorber solution storage bottles.

8.2.4 Leak test

A leak test shall be carried out before and after each sample run and after any point when the system has been disassembled during the sample period.

8.2.4.1 Method A

No additional requirements

8.2.4.2 Method B

No additional requirements

8.2.4.3 Method C

No additional requirements

8.2.5 Installation of equipment at chimney or duct

No additional requirements

8.3 Performance of the sampling

No additional requirements
8.4 Disassembling the equipment
It may be necessary to allow the sampling equipment to cool before disassembly. Under these circumstances care shall be taken to prevent contamination via the nozzle or other exposed opening. Aluminium foil shall not be used to cap sample trains.

8.4.1 Disassembling of the filter housing
No additional requirements

8.4.2 Rinsing of the sampling equipment
No additional requirements

8.4.3 Rinsing of the connection tubing to the first absorber
No additional requirements

8.4.4 Handling of the absorption solutions from the absorbers
No additional requirements

8.5 Field blanks
A blank shall be taken at each stack during a sampling campaign. A campaign is defined as one site visit.

When carrying out a sampling campaign with multiple tests over a number of days it is advisable to carry out several blanks.

When measuring mercury it is necessary to carry out analysis of mercury upstream of the mercury absorbers. This analysis shall also form part of the field blank.

The blank procedure shall include carrying out a leak check following the requirements of 8.2.4.

8.6 Requirements for storage of the samples
Samples shall be stored during transport in a cool dark place, such as a cool box packed with ice or plastic freezer packs. An appropriate temperature measurement / indication device shall be used to confirm that the temperature is maintained below 25°C.

Note: this approach is based on the requirements in US EPA M29 and VDI 3869.

8.7 Pre-treatment before analysis
No additional requirements

8.7.1 General
No additional requirements

8.7.2 Pre-cleaning of the digestion equipment
No additional requirements

8.7.3 Treatment of CRM and filter
No additional requirements

8.7.3.1 General procedure
No additional requirements

8.7.3.2 Heating in oven or heating plate
No additional requirements
8.7.3.3 Heating in microwave oven

When analysing metals including mercury the requirements of EN 13211 shall be met.

Note: EN 13211 specifies that the filter should be heated during digestion for a greater length of time than specified in EN 14385.

8.7.4 Pre-treatment of absorption solutions

No additional requirements.

8.7.5 Pre-treatment of rinsing solutions

The pre-treatment procedure of rinsing solutions (used for rinsing sampling equipment, such as nozzle, probe and filter housing) prevents the metals method being combined with particulate sampling.

When measuring mercury it is necessary to carry out the pre-treatment according to EN 14385. However, before carrying pre-treatment for metals analysis a portion shall be quantitatively removed for mercury analysis.

8.8 Analysis

To analyse mercury, EN13211 specifies the use of EN 1483 Water Quality - Determination of mercury. It is recognised that analytical methods change as new techniques and equipment are developed. It is therefore acceptable to use different procedures provided it is demonstrated that they meet the performance characteristics of EN1483.

When measuring mercury it is necessary to meet the requirements of EN 13211. For the filter and probe wash the following approach shall be used:

- the filter shall be digested/heated and analysed as one sample following the requirements in BS EN 14385. This gives the result for both the mercury and metals on the filter;
- the probe rinse volume shall be measured. This shall be split into two samples A and B (B being a measured small aliquot);
- Sample A shall be digested/heated and analysed according to BS EN 14385. This gives the result for metals in the sample A portion of the probe rinse. The metals concentration in sample A shall be used to determine the metals mass in sample B. The results from samples A and B shall be combined to give the total metals in the probe rinse.
- Sample B shall be analysed according to BS EN 13211. This gives the probe rinse mercury result for sample B. The mercury concentration in B shall be used to determine the mercury mass in Sample A. The results from samples A and B shall be combined to give the total mercury in the probe rinse.

Besides these requirements it is also necessary to analyse the absorber solutions in the first three absorbers, which are primarily in place to capture other gas phase metals. In total five absorbers shall be measured for mercury by taking a known volume from each. The analytical method for measuring all five absorbers is the same. At least once per sampling campaign it shall be proven that less than 5% of the total gas phase mercury absorbed is in the final absorber (see section 7.33 of EN 13211).

The analysis of the blank shall be identical to the analysis of the sample (if more than one sample is taken at the stack during the sampling campaign, the blank shall be identical to the sample used to determine the capture efficiency).
If the standards are not matrix-matched then care should be taken to ensure that there are no interference effects present.

Linearity shall be checked and the correlation coefficient should be 0.98 or better. Analyte additions can be used as an alternative calibration technique and internal standards may be added to correct for transport effects and instrument drift.

Suitable quality checks (QCs) should be prepared and measured with each batch of samples. For filters, a suitable certified reference material (CRM) can be used if available. If no CRMs are available then a spiked filter may be exposed to the digestion process instead. For adsorption and rinsing solutions, spiked solutions can be used. For all QCs a recovery of 100 ± 20 % of the expected or certified value is acceptable. If any QC results are outside this range then this must be highlighted in the final report. If recoveries are outside the required range sampling and analysis stages may have to be repeated.

Note: adsorption and rinsing solutions are tested as part of the blank solution for digestion and, consequently, do not need to be done separately.

9. Expression of results

If results are at or below the limit of detection then they shall be reported as being at the detection limit.

10. Test report

If carrying out MCERTS accredited work the monitoring report shall be submitted in the report format specified in the Manual stack emission monitoring performance standard for organisations (Annex H) available from www.mcerts.net. The report shall also include specific requirements required by EN14385 and this MID.
Annex A: Examples of absorption vessels
Example B Greenburg-Smith impingers shall be used.

Annex B: Types of isokinetic equipment
The examples shown below are shown with out of stack filtration.

Figure 1: Sample train for metals (excluding mercury)

Absorber 1 to 3 absorption solution (see 5.2.8)
Absorber 4 empty
Absorber 5 Desiccant (e.g Silica Gel)
Annex C: Pre-cleaning procedures of the sampling equipment at the laboratory and determination of the absorption efficiency

The cleaning procedure given in section 7 of this MID shall be used.

Annex D: Measurement results of two field tests

No additional information

Annex E: pre-tests for determination of the efficiency, of the digestion and of the performance of the analytical procedure

No additional information

Annex ZA: Relationship with EU Directives

No additional information

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